



Bulk Metallic Glasses - Characteristics and Possibilities

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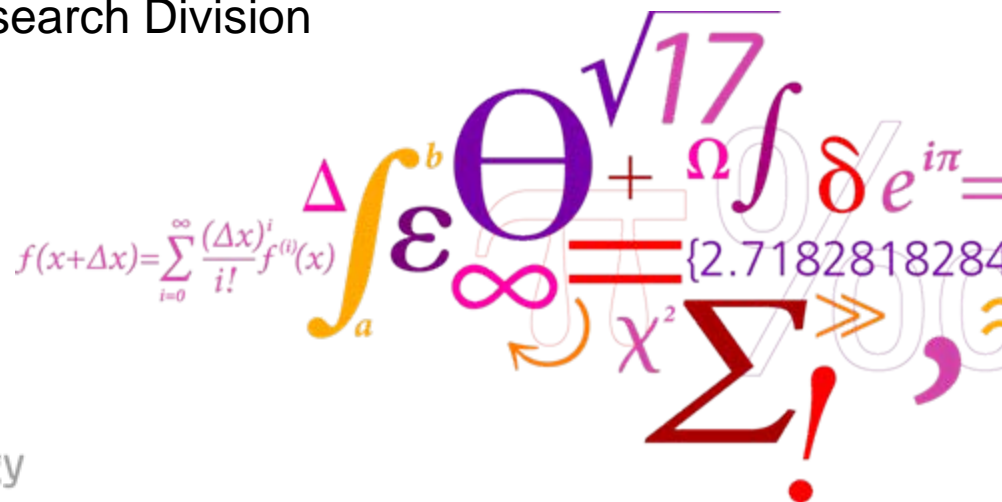
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Bulk Metallic Glasses – Characteristics and Possibilities

Morten Eldrup and Cormac Byrne
Materials Research Division



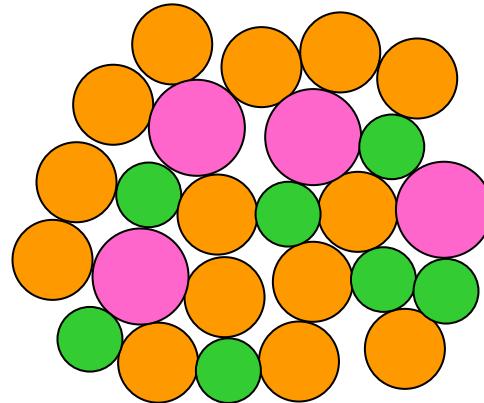
Overview

- What is a metallic glass
- How to make a metallic glass and a
- Bulk Metallic Glass (BMG)
- Some properties of BMGs
 - Supercooled liquid range
 - Surface imprinting
 - Forming in 3D
 - Problems
 - Near-net-shaping
 - Mechanical properties
- Conclusion

What is an amorphous alloy/metallic glass?

Glass

Solid where the atom positions are disordered



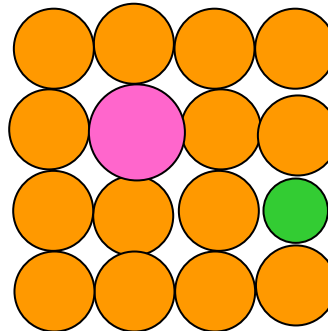
Ceramic

Polymer

Metallic

Crystal

Solid where the atom positions are ordered



Ceramic

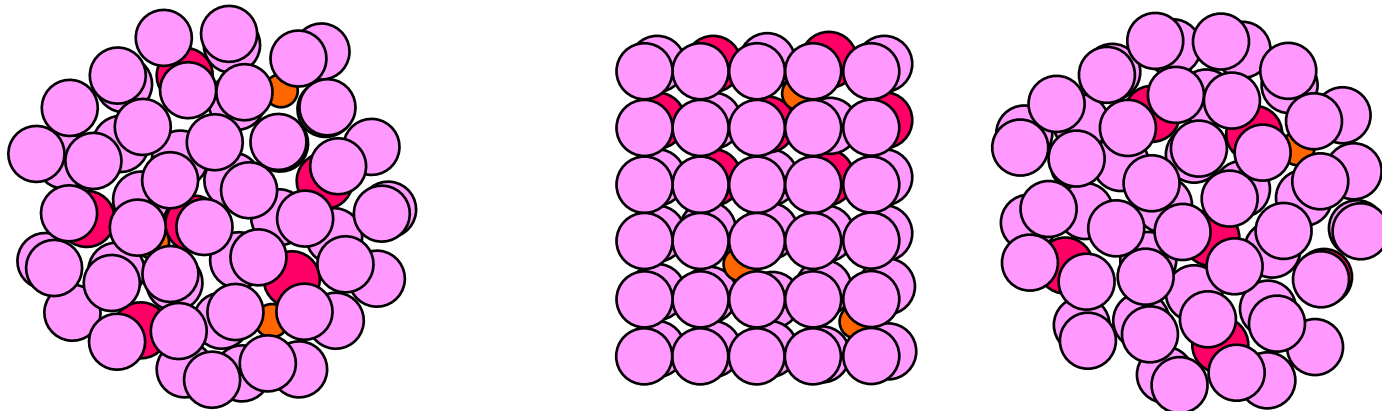
Polymer

Metallic

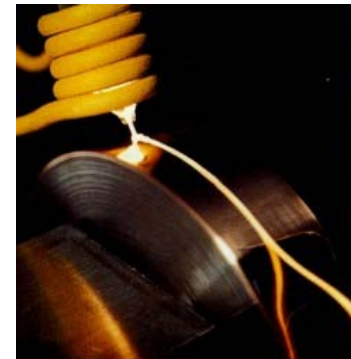
Discovery of metallic glasses

- 1959 First report of rapid cooling of a metallic alloy to form a glass
Paul Duwez, California Institute of Technology, USA

Metallic glass



Rapid cooling $\sim 10^6$ K/s
Results in thin ribbons



Discovery of metallic glasses

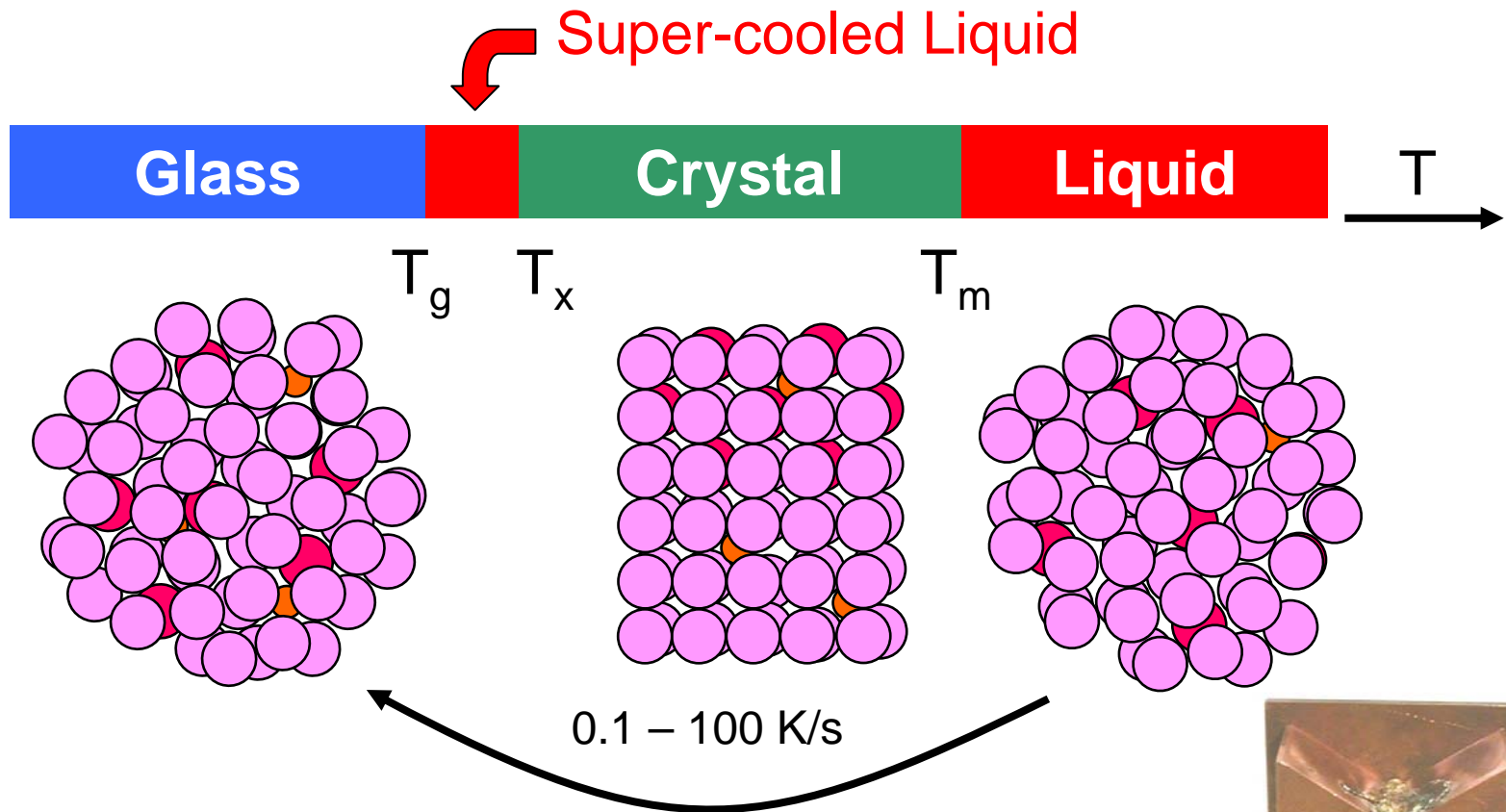
- 1959 First report of rapid cooling of a metallic alloy to form a glass
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- 1970s Magnetic applications of rapidly cooled metallic glasses
Electrical transformers, antitheft strips

Discovery of metallic glasses

- 1959 First report of rapid cooling of a metallic alloy to form a glass
Paul Duwez, California Institute of Technology, USA
- 1970s Magnetic applications of rapidly cooled metallic glasses
Electrical transformers, antitheft strips
- 1984 First report of metallic alloys that avoid crystallization without rapid cooling by Kui et al., Harvard University, and Akihisa Inoue et al., Tohoku University, Japan (1989)

Bulk metallic glass = BMG

Bulk Metallic Glass



Moderate cooling rate is
sufficient to avoid crystallisation



MIKROMETAL

Making BMG Castings

Drop casting in arc melter
~4 g

Rods 3 - 5 mm Ø, 25 mm long

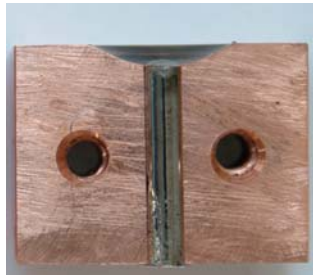
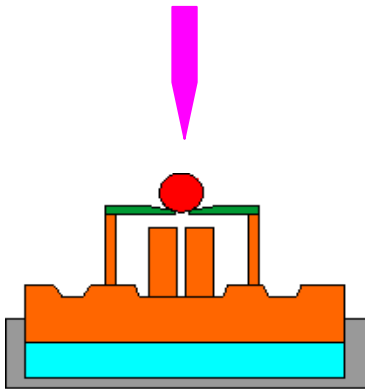
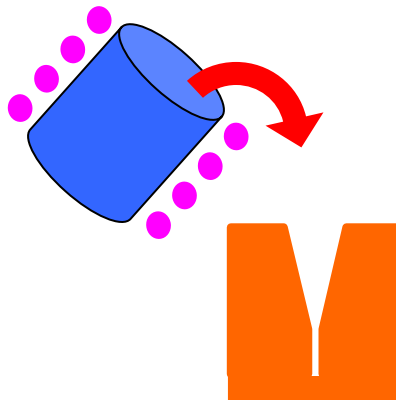


Plate casting in induction melter ~ 10 g

25 x 25 x 2 mm³ plates



Ribbon Spinning ~ 3 g

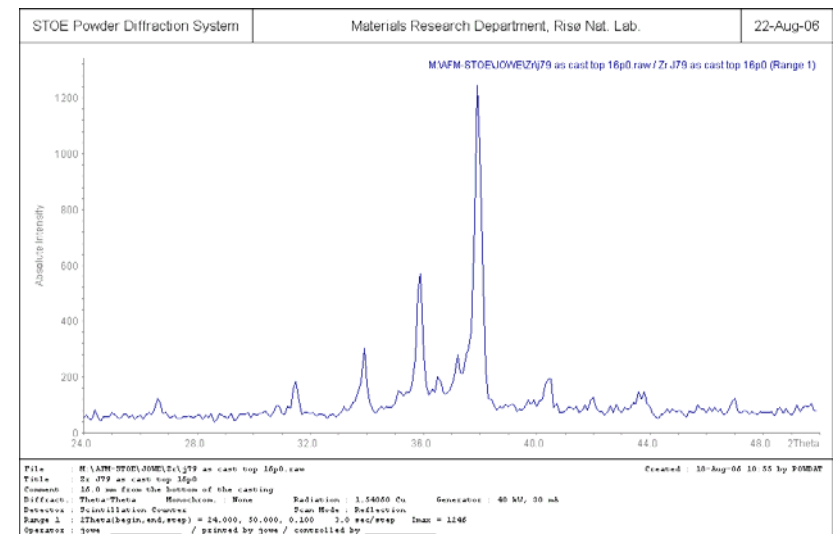
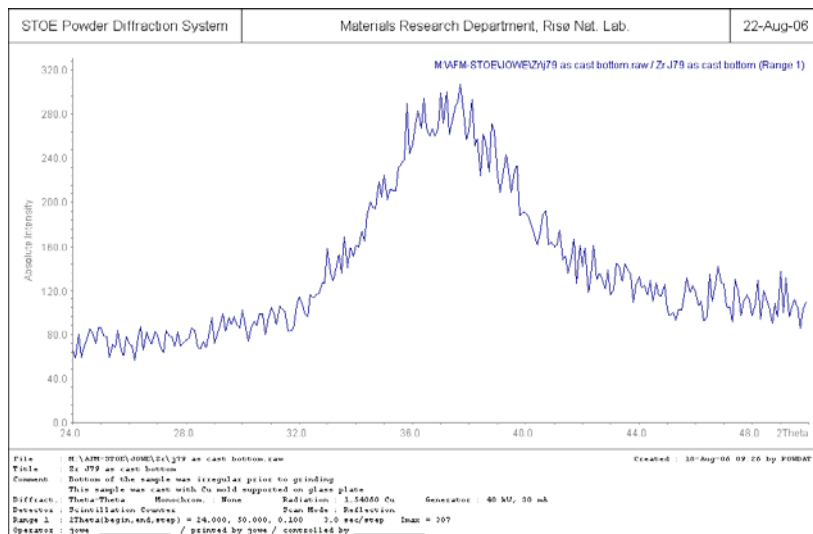
70 µm x 4 mm x 3 m



Specimens produced by melt spinning,
drop casting or suction casting



- No visual difference between amorphous and crystalline casting
- Check castings by x-ray

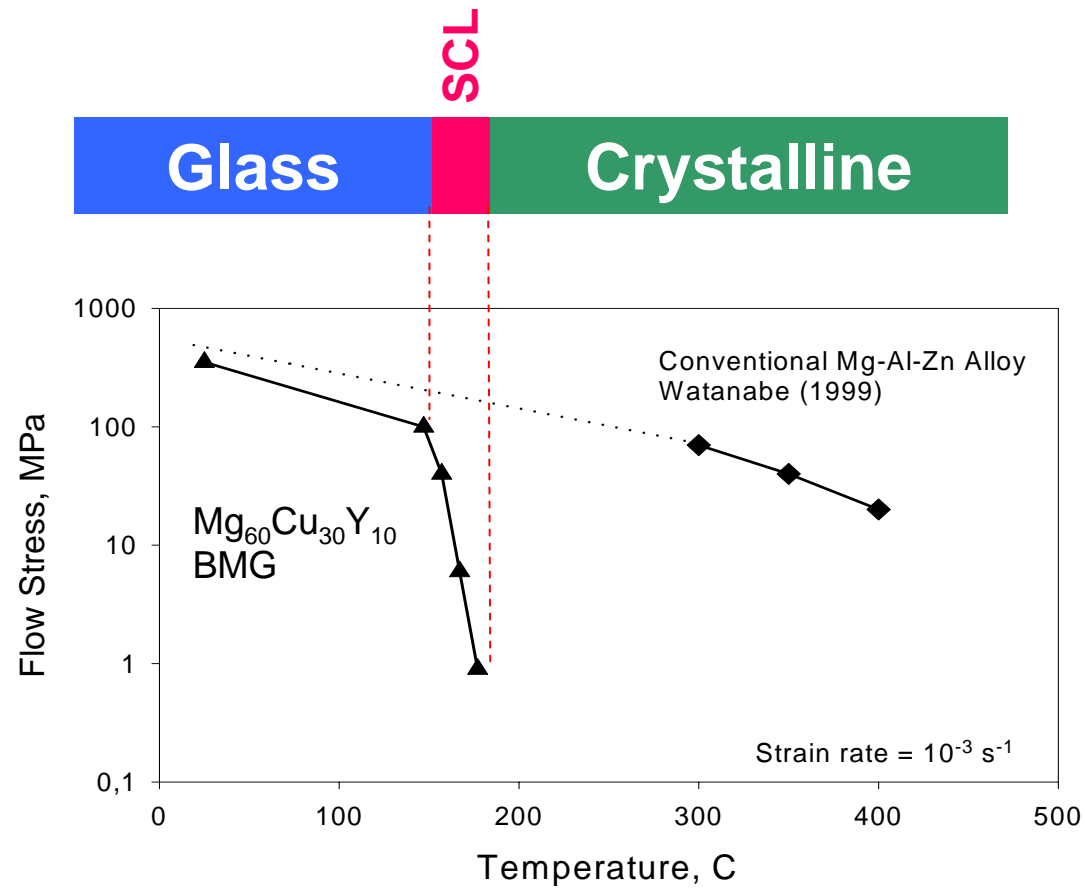


Examples of BMG

Alloy Family	Specific alloy	Critical cooling rate (K/s)	T _g (K)	T _x (K)
Ti-Zr-TM	Ti ₃₄ Zr ₁₁ Cu ₄₇ Ni ₈	250	698	727
Mg-Ln-(Cu,Ni, Zn)	Mg ₆₅ Cu ₂₅ Y ₁₀	50	425	479
Ln-Al-TM	La ₅₅ Al ₂₅ Ni ₁₅ Cu ₅	35	474	541
Zr-Ti-Al-TM	Zr _{52.5} Ti ₅ Al ₁₀ Ni _{14.6} Cu _{17.9}	20	~ 700	~ 795
Zr-Ti-TM-Be	Zr _{38.5} Ti _{16.5} Ni _{9.75} Cu _{15.25} Be ₂₀	2	630	678
Pd-Cu-Ni-P	Pd _{42.5} Cu ₃₀ Ni _{7.5} P ₂₀	0.067	574	660

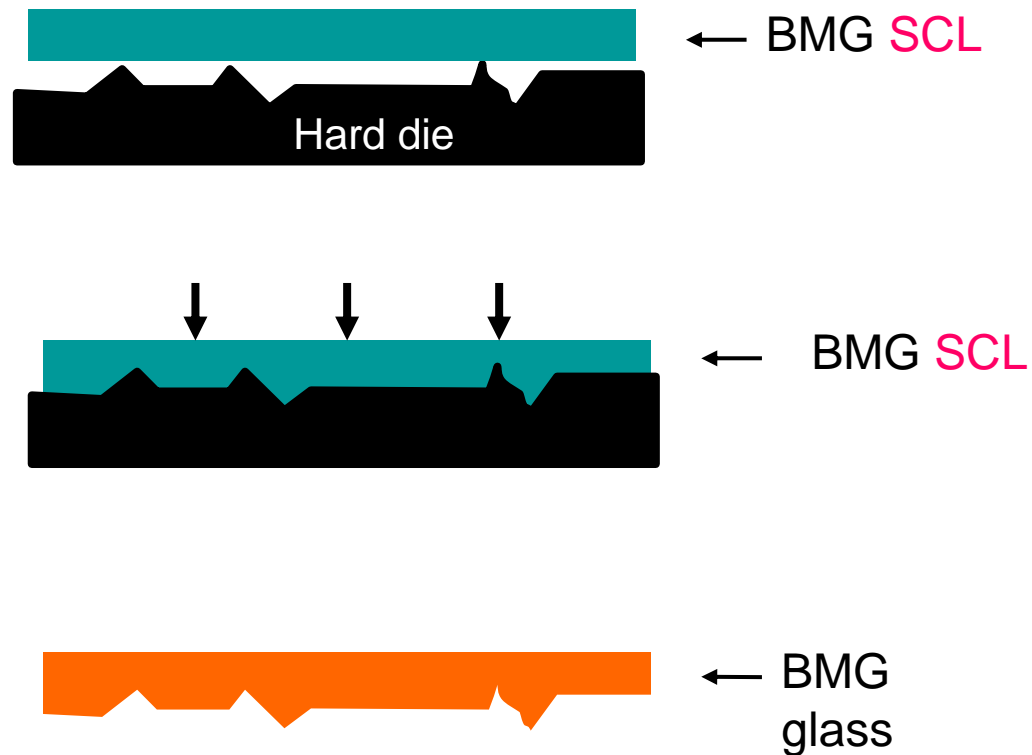
TM = Group VI–VIII transition metals (Cr, Mo, W, Mn, Re, Co, Rh, Ir, Ni, Pd, Pt)

Ln = Y, Lanthanide metals (La or rare earth metals)



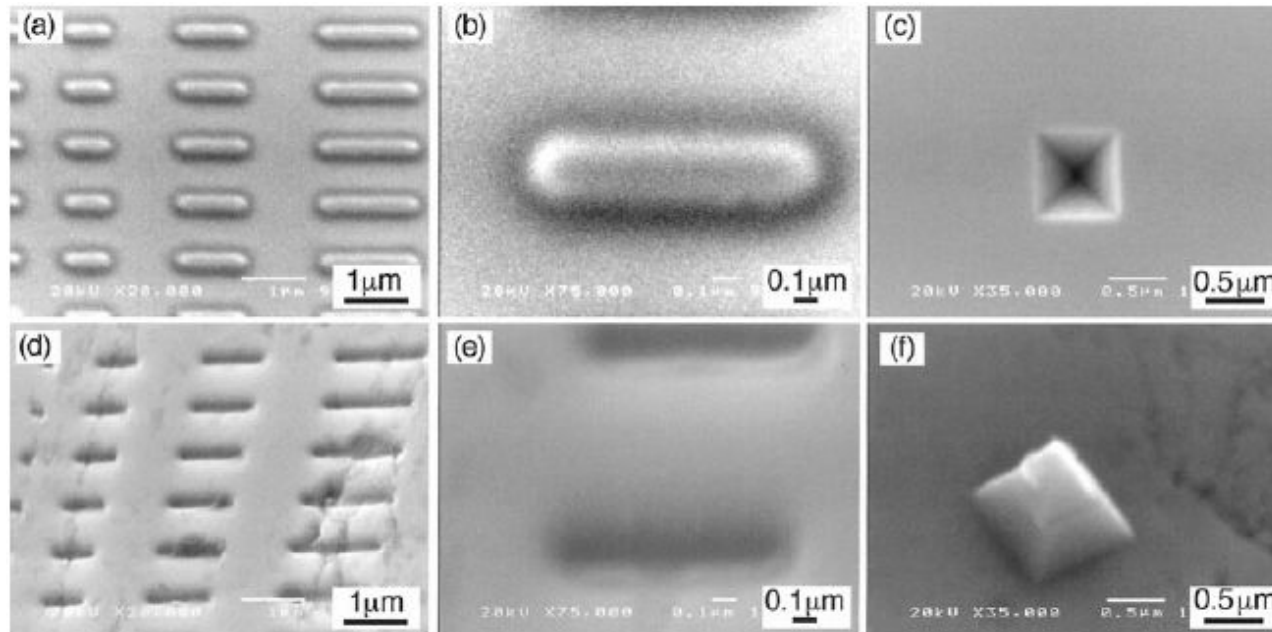
Shaping Surfaces

with micro- or nano-scale topology

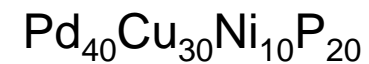


Some patterns

Si/SiO₂
master

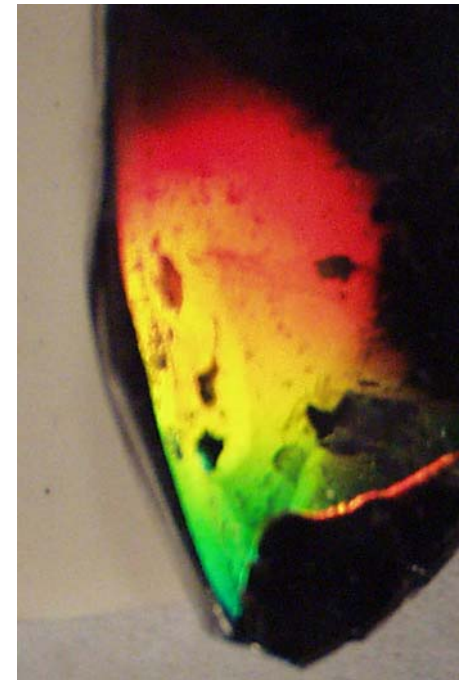
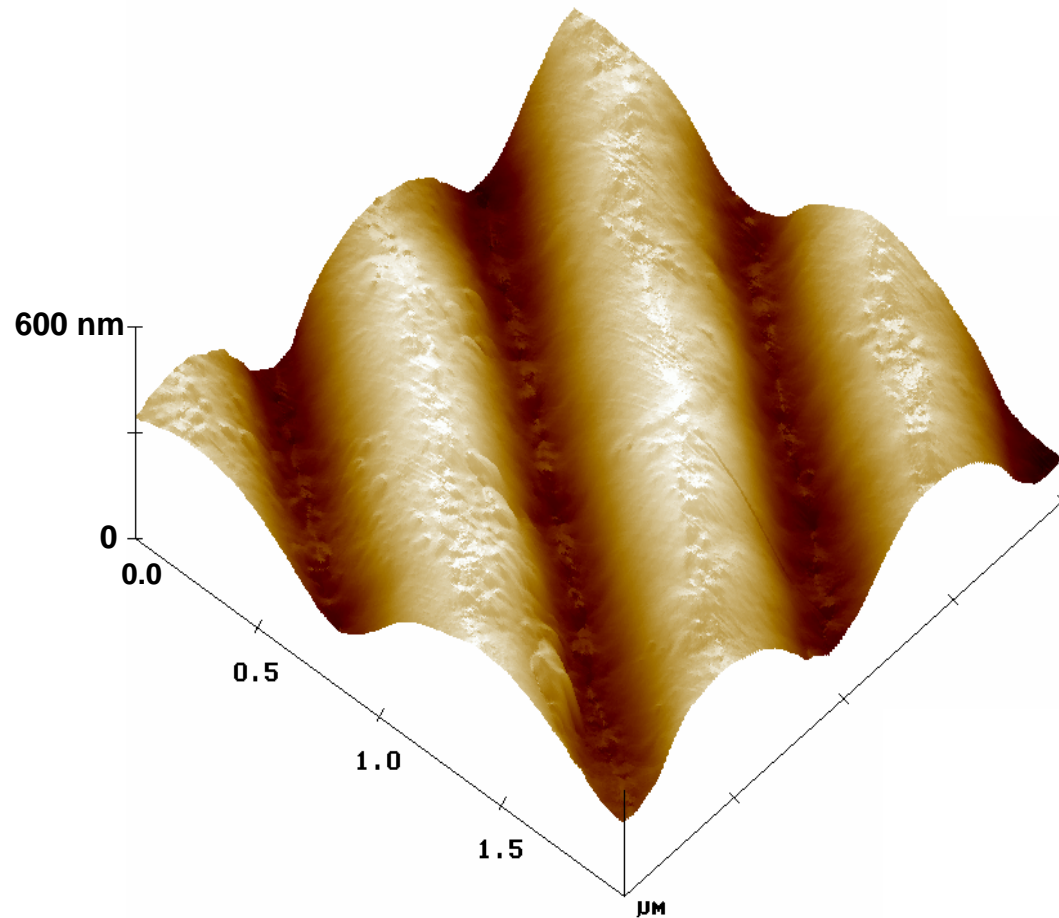


BMG
replica



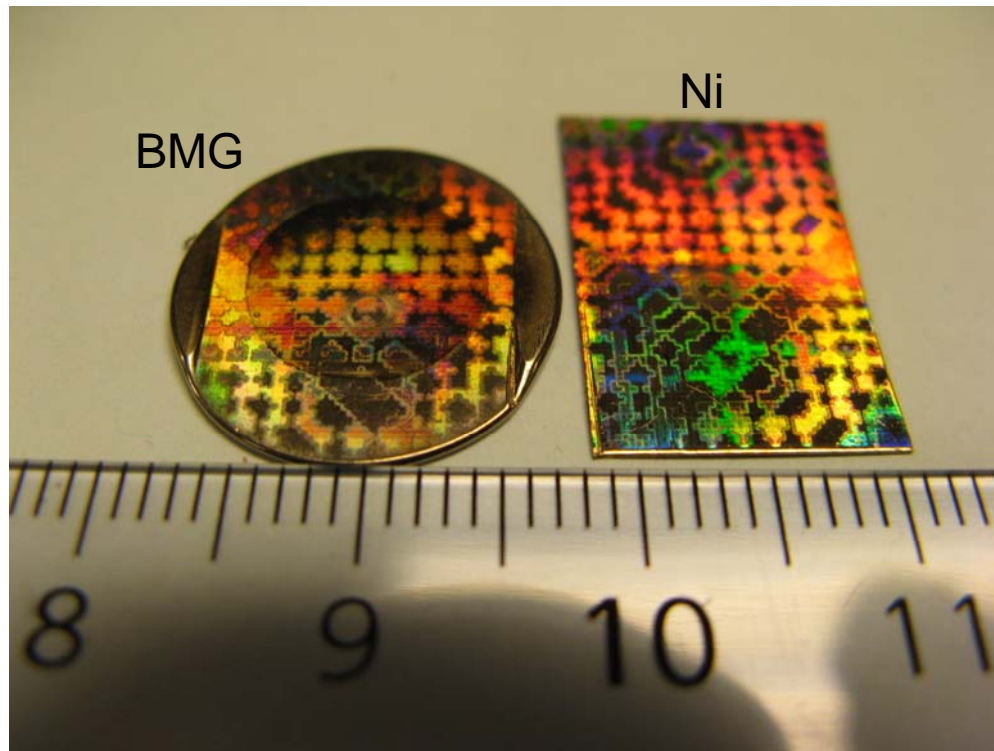
$T = 640\text{K}$, $P = 10\text{ Mpa}$, time = 1000s

Linear sine pattern replicated on $\text{Mg}_{60}\text{Cu}_{30}\text{Y}_{10}$

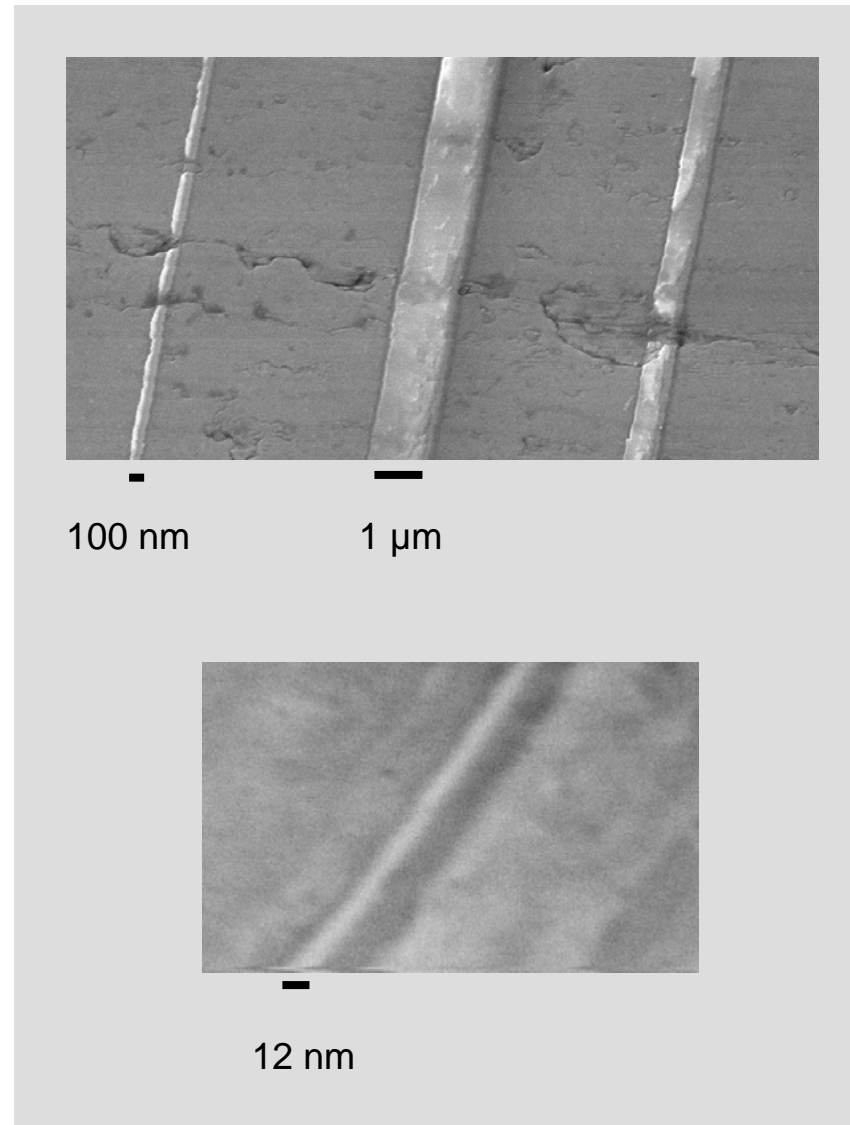


How small can these features be?

Diffraction pattern on Ni surface imprinted on a BMG ($\text{Zr}_{44}\text{Ti}_{11}\text{Cu}_{10}\text{Ni}_{10}\text{Be}_{25}$)



Nanoscale surface topology



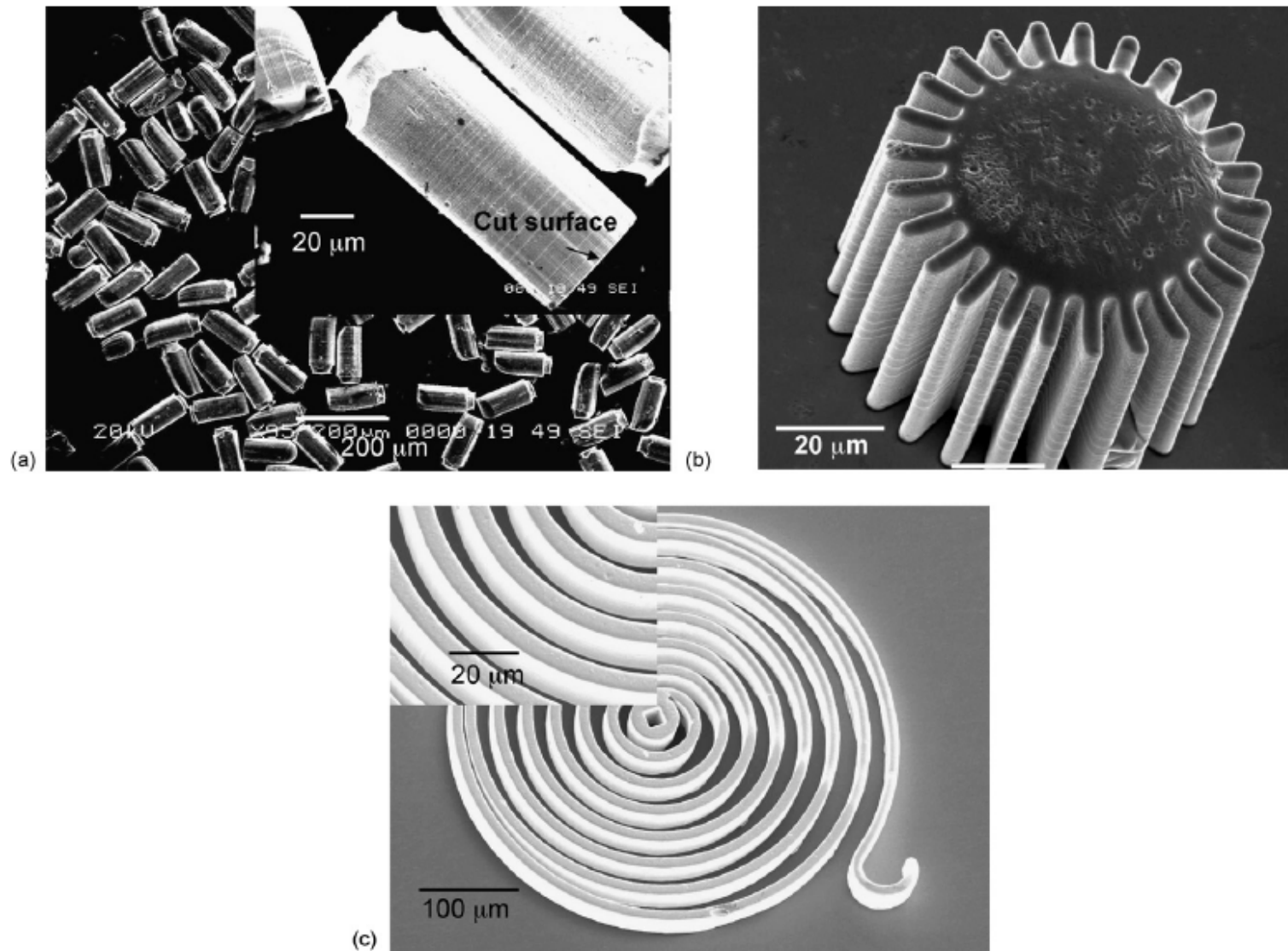
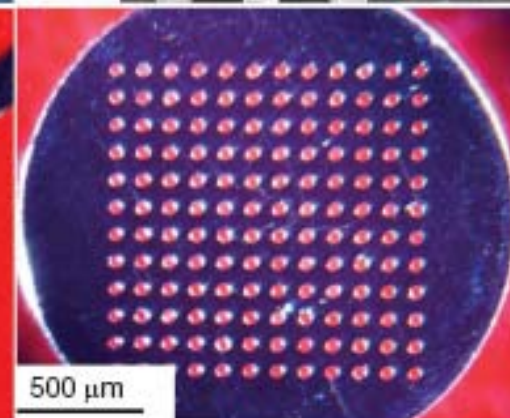
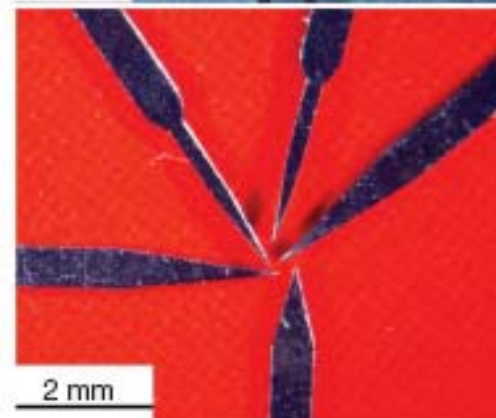
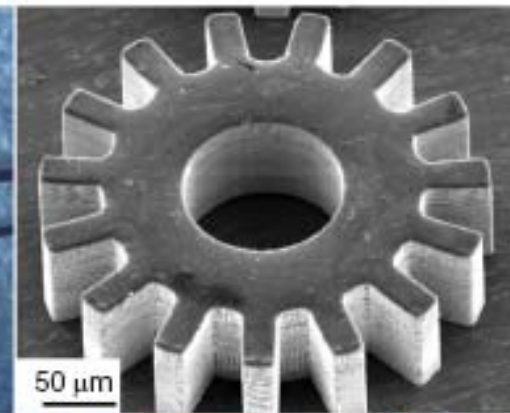
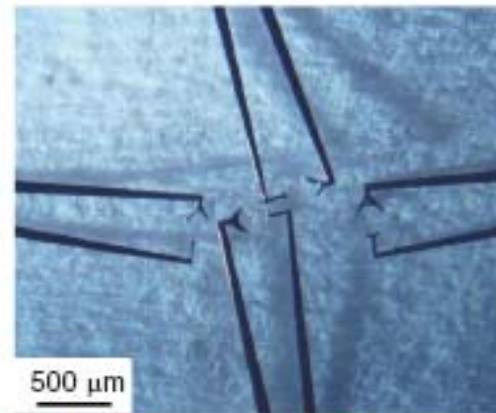
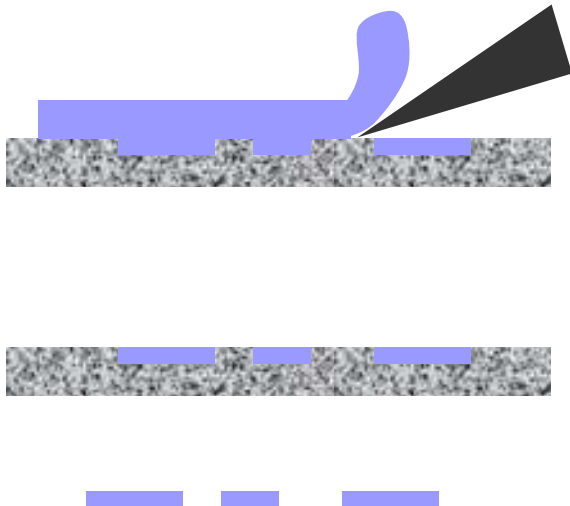
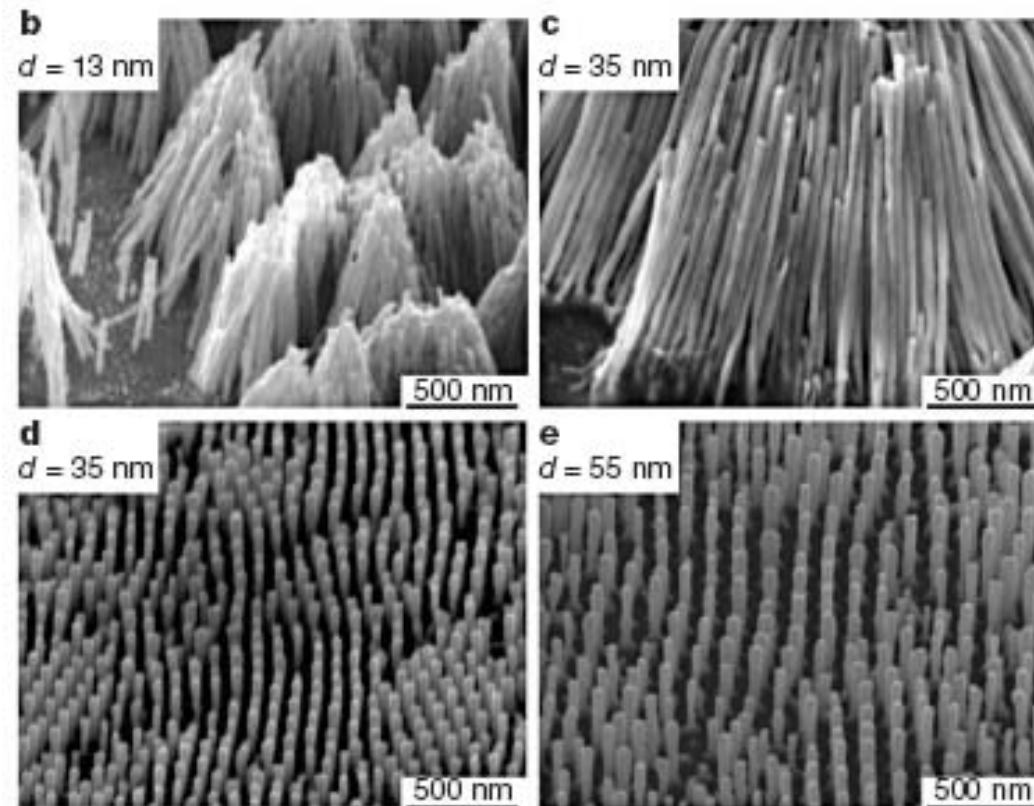


Fig. 4. Three-dimensional parts obtained by using TPF of BMG and subsequent hot-separation method. (a) Rods where the smooth cutting surface is visible. (b) Complex micro gears. (c) Coil shape spring.

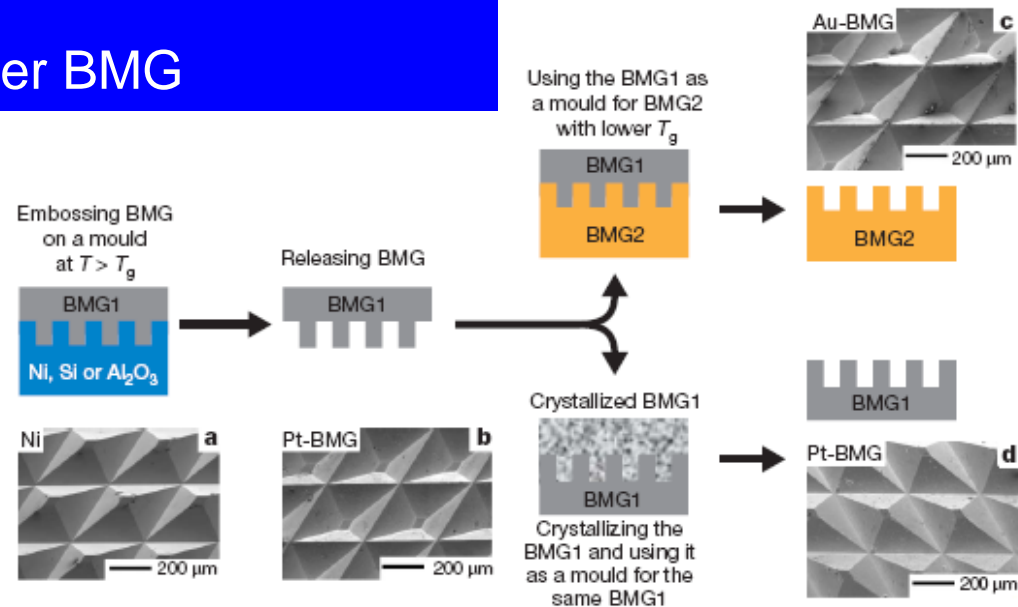
HOT CUTTING OF BMGs



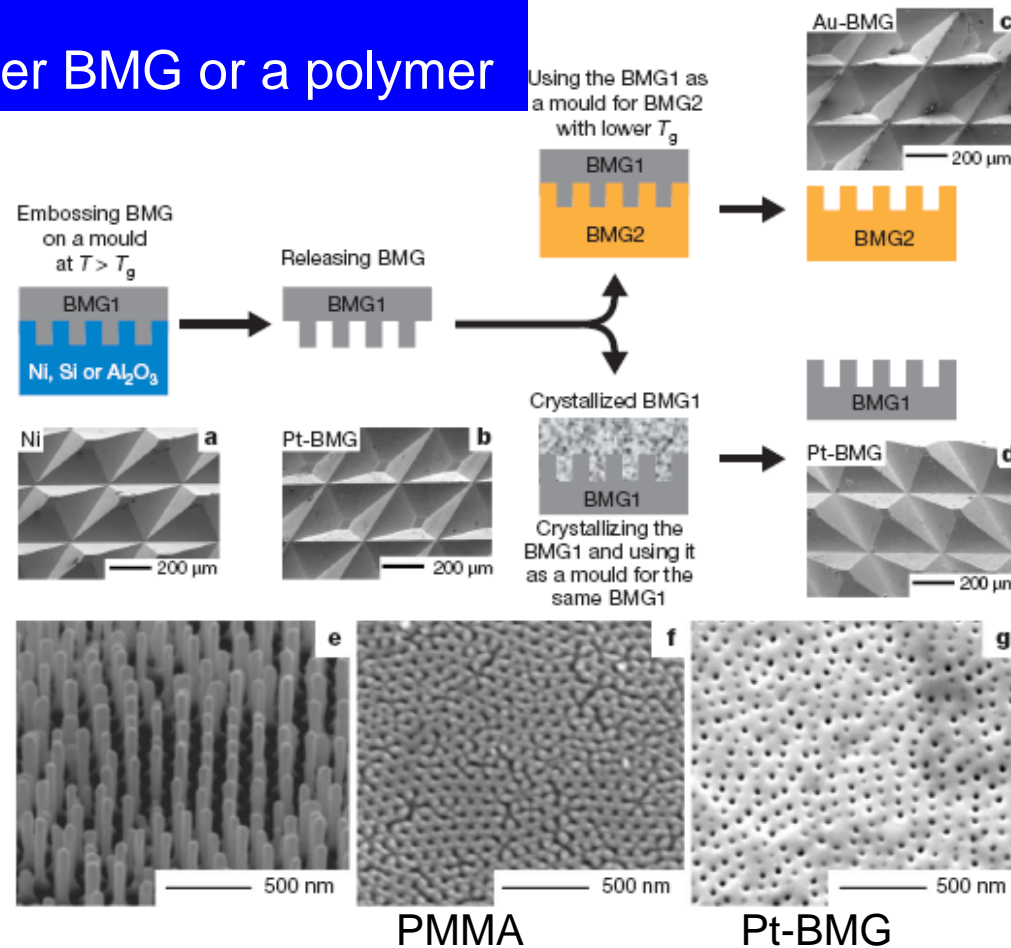
Pt-BMG rods formed by embossing on porous alumina

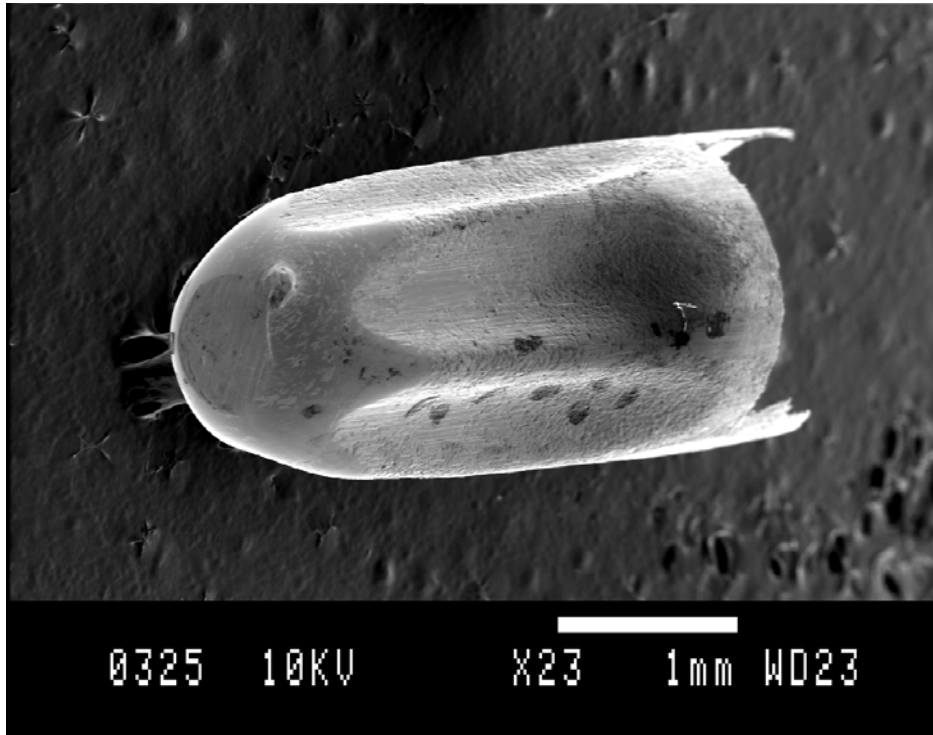


Using one BMG as a mould for another BMG

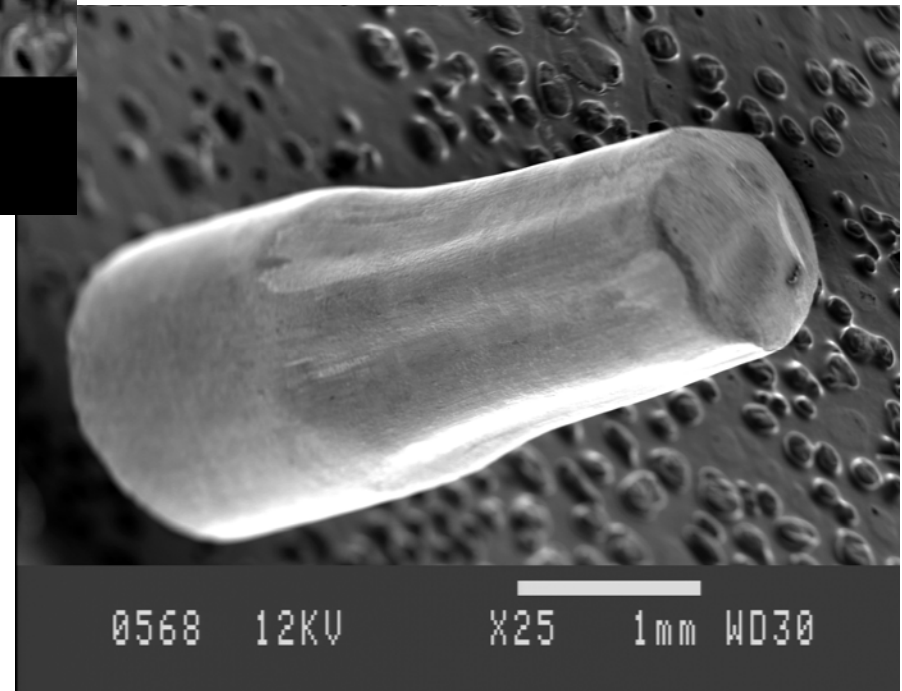


Using one BMG as a mould for another BMG or a polymer





BMG screwdriver tip



BMG-material:

Vitreloy 1B, Vit1B, LM1B

$(\text{Zr}_{44}\text{Ti}_{11}\text{Cu}_{10}\text{Ni}_{10}\text{Be}_{25})$

CTE $\sim 10 \times 10^{-6}/\text{K}$

$T_g \sim 350^\circ\text{C}$

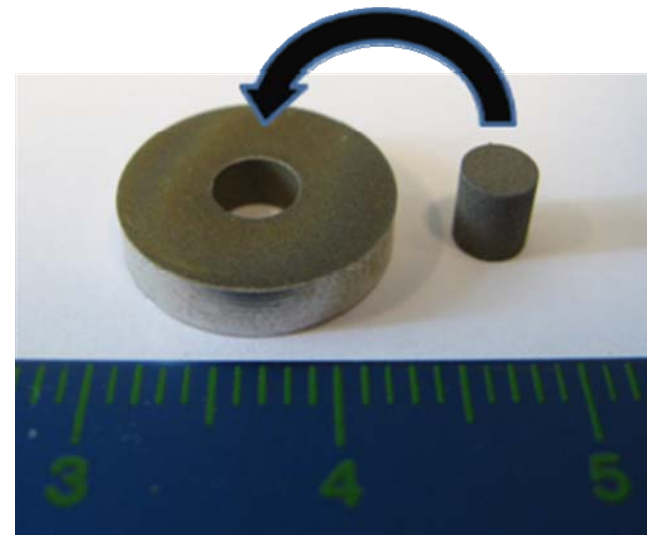
$T_x \sim 470^\circ\text{C}$

Dies with different thermal expansion coefficients:

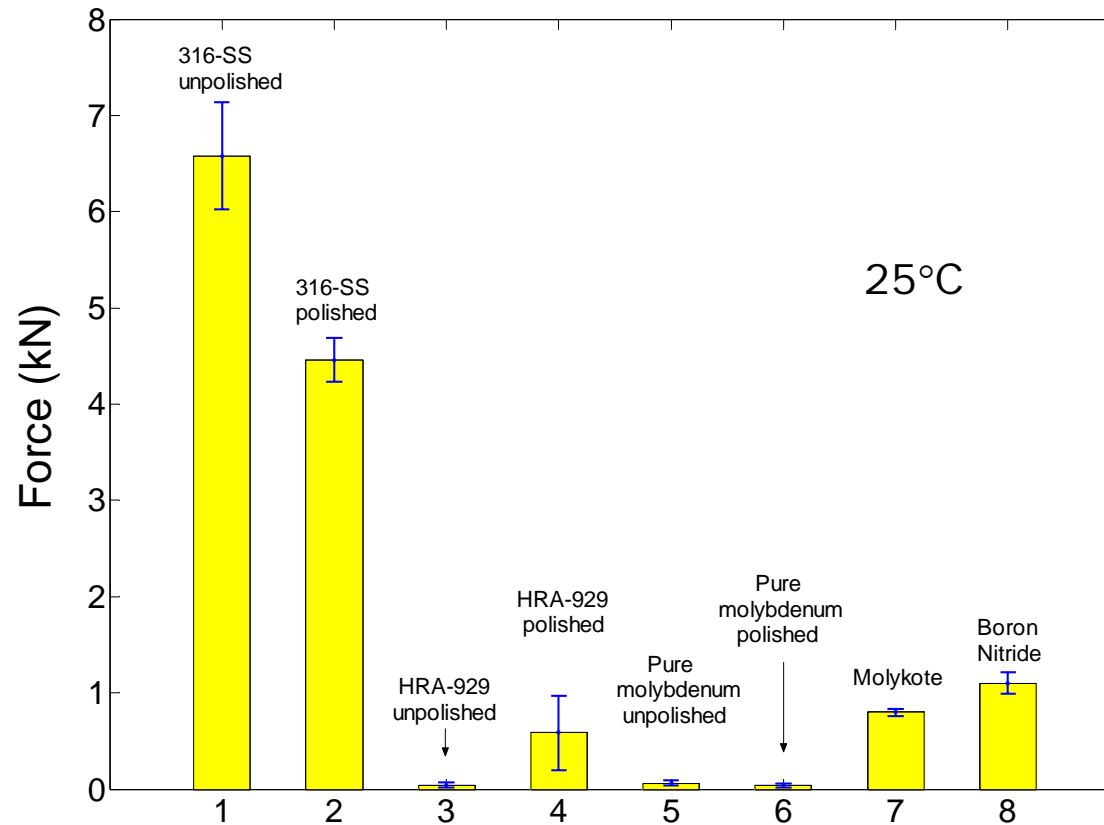
- Stainless steel (CTE $\sim 18 \times 10^{-6}/\text{K}$)
- Pure molybdenum (CTE $\sim 5 \times 10^{-6}/\text{K}$)
- HRA-929 superalloy (CTE $\sim 6 \times 10^{-6}/\text{K}$) (Hitachi)

Polished and un-polished

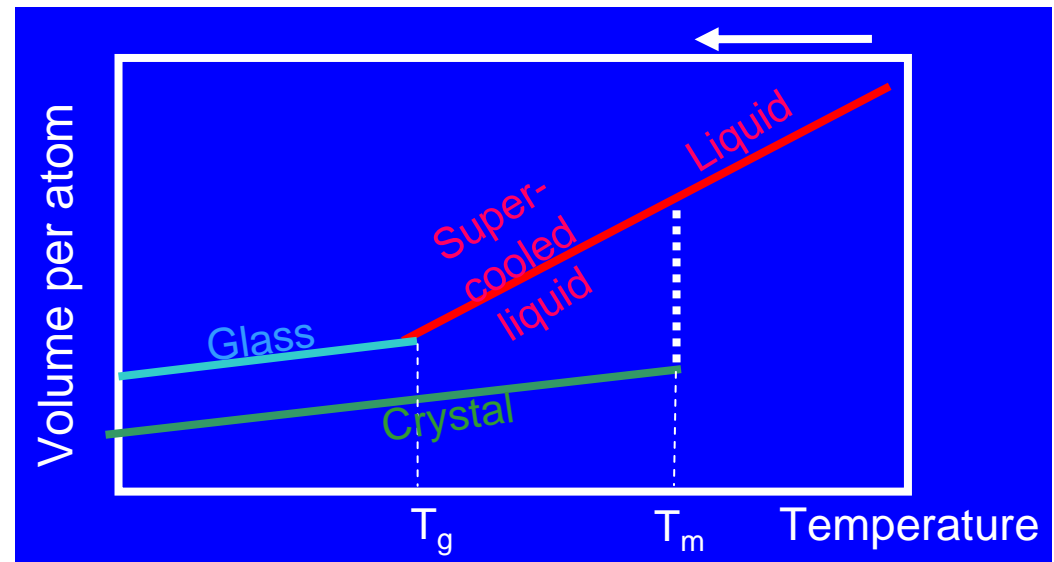
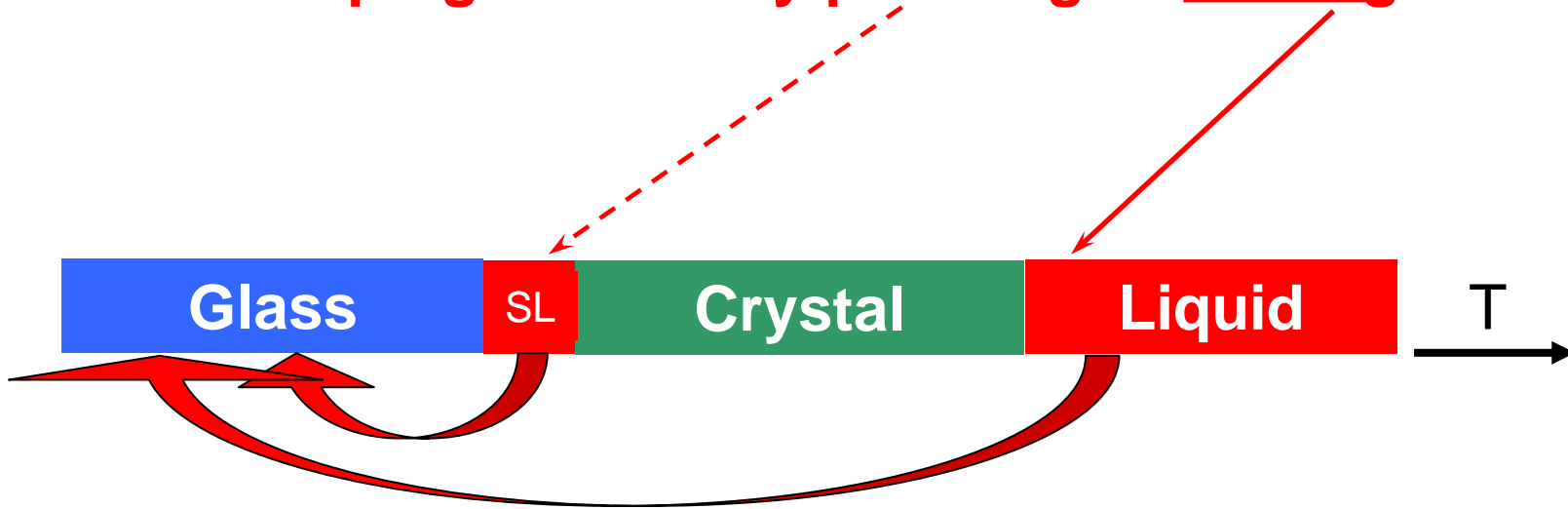
Ejection Experiment



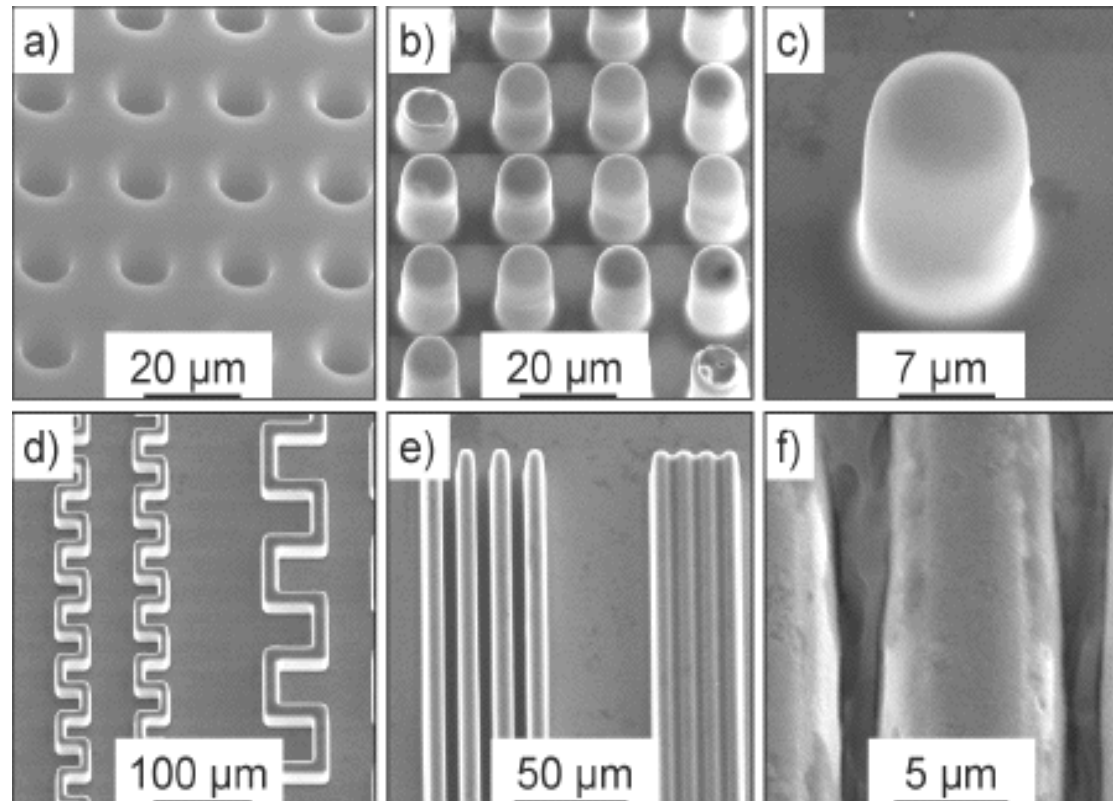
Maximum ejection force



Shaping of BMG by pressing or casting

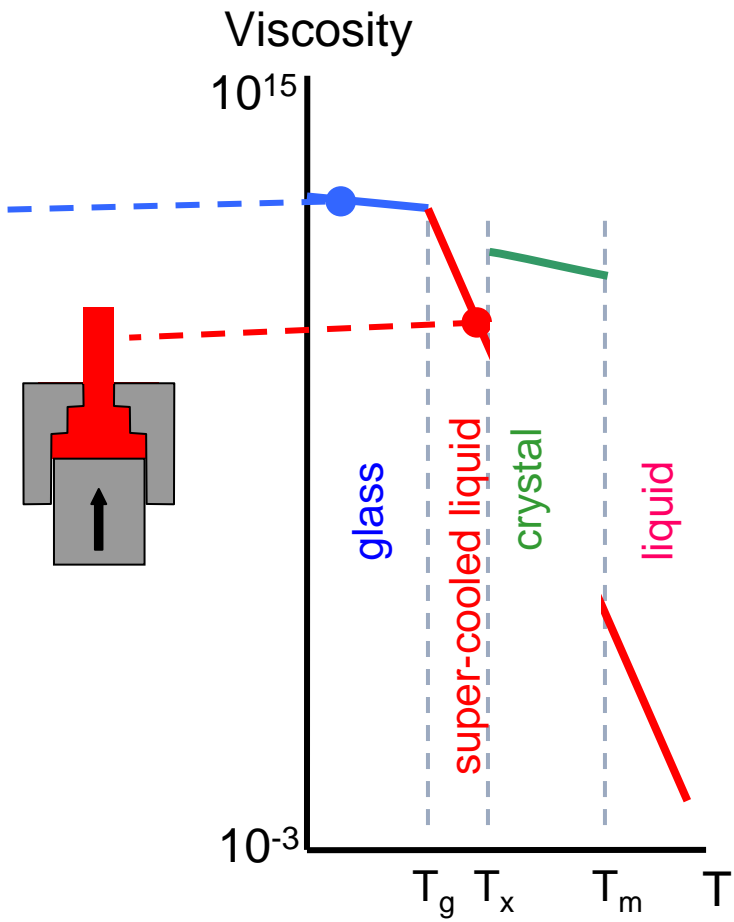
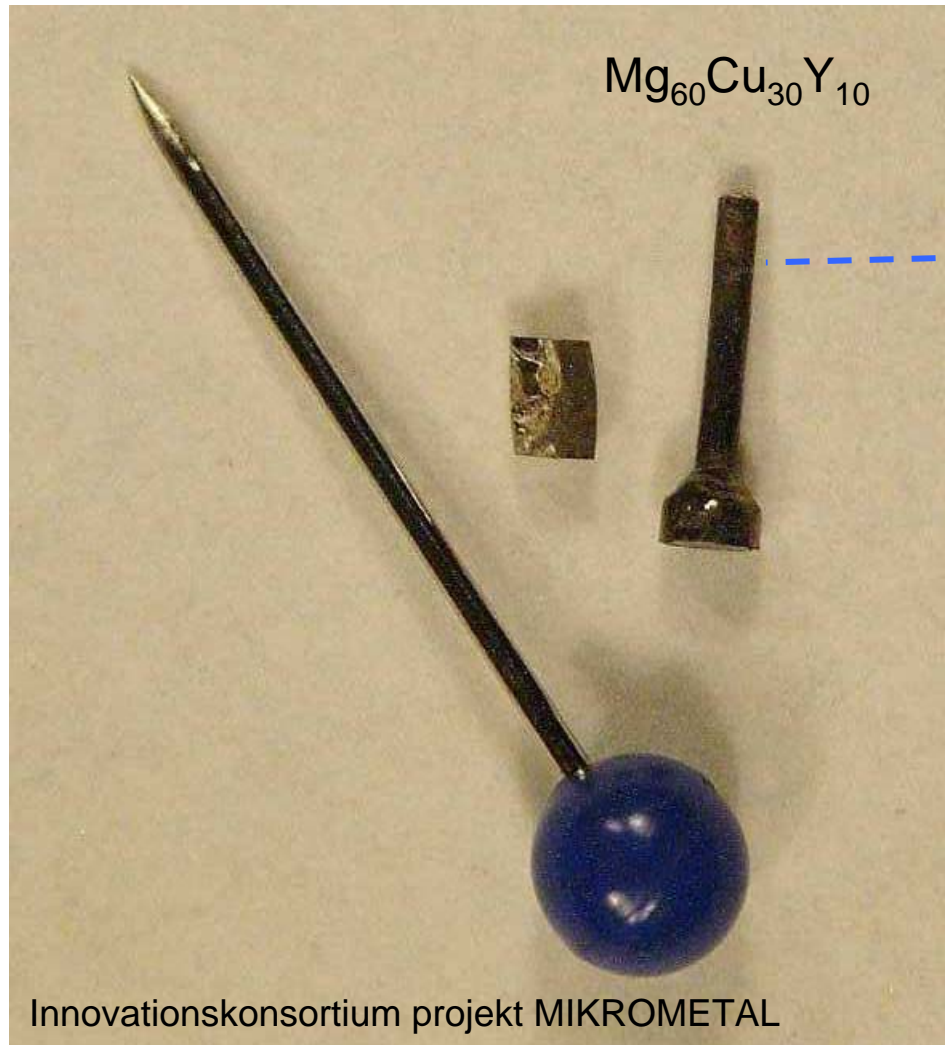


Casting



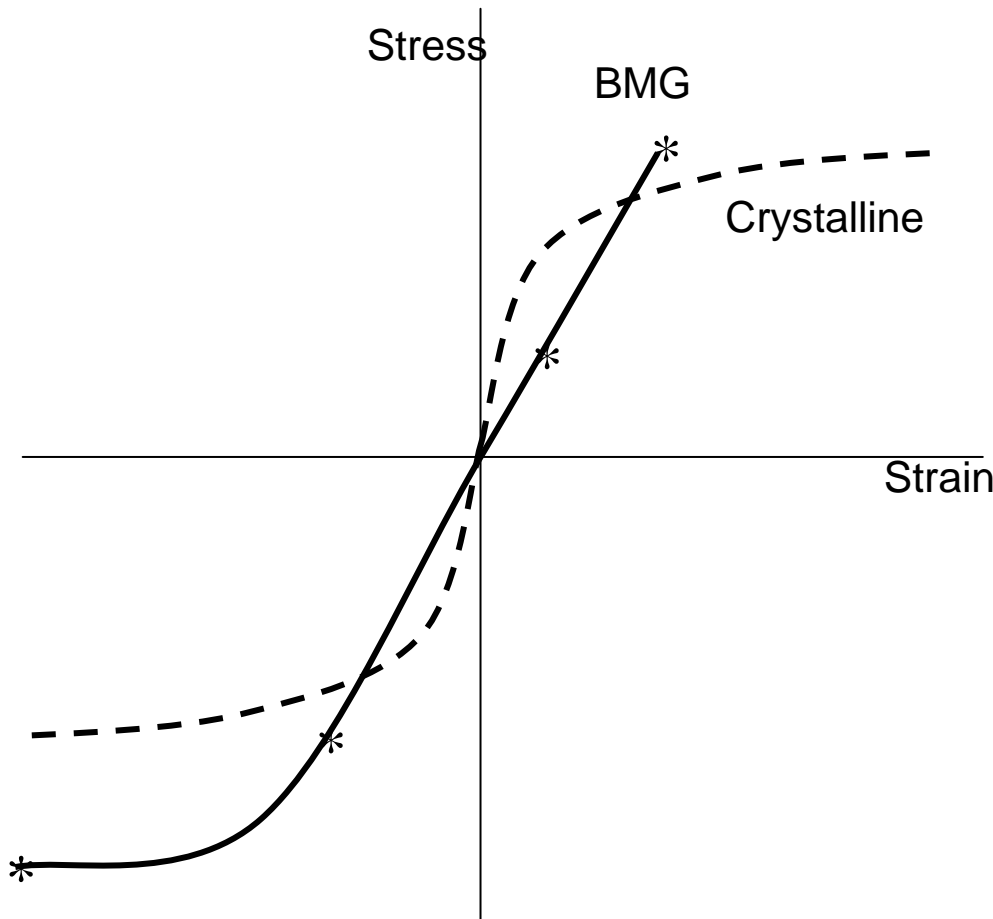
- to near net-shape

Extruded microcomponent (10 mm – 10 μm)



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Characteristic BMG properties below T_g



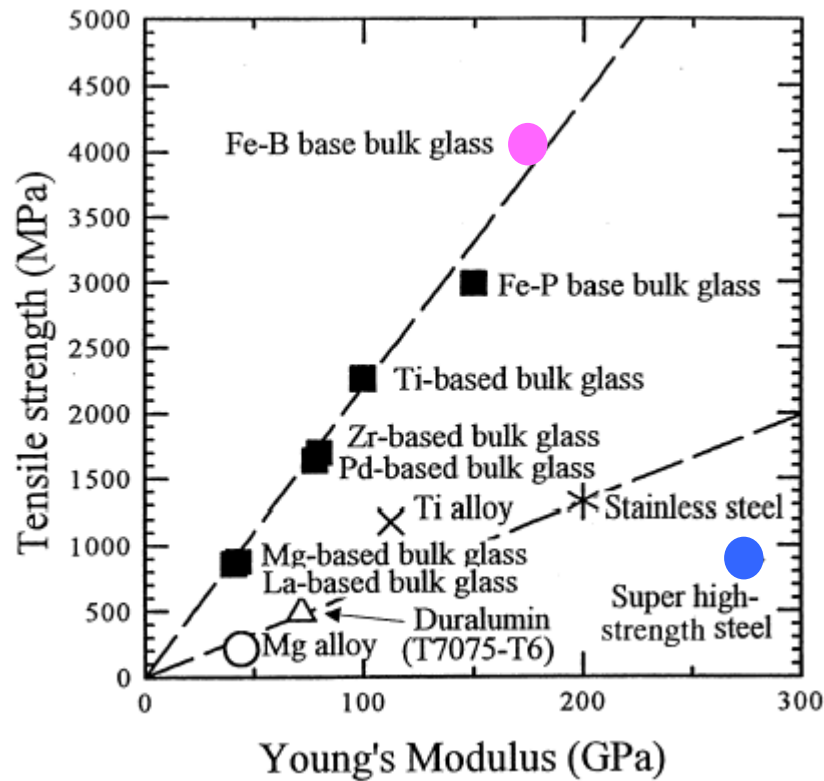
Modulus ca. 40% of
cryst. alloy

Virtually no plastic flow
in tension

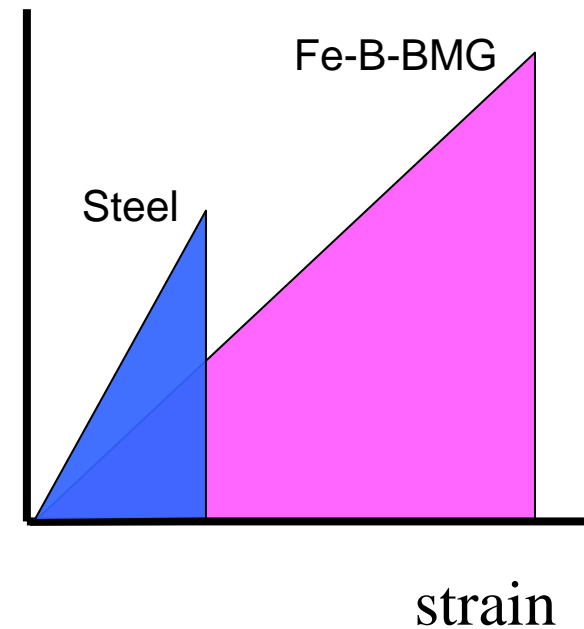
Fracture stress in
tension (*) depends on
BMG and quality of
casting

0 - several % plastic
flow in compression

Elastic Modulus



stress



Energy Storage Comparison

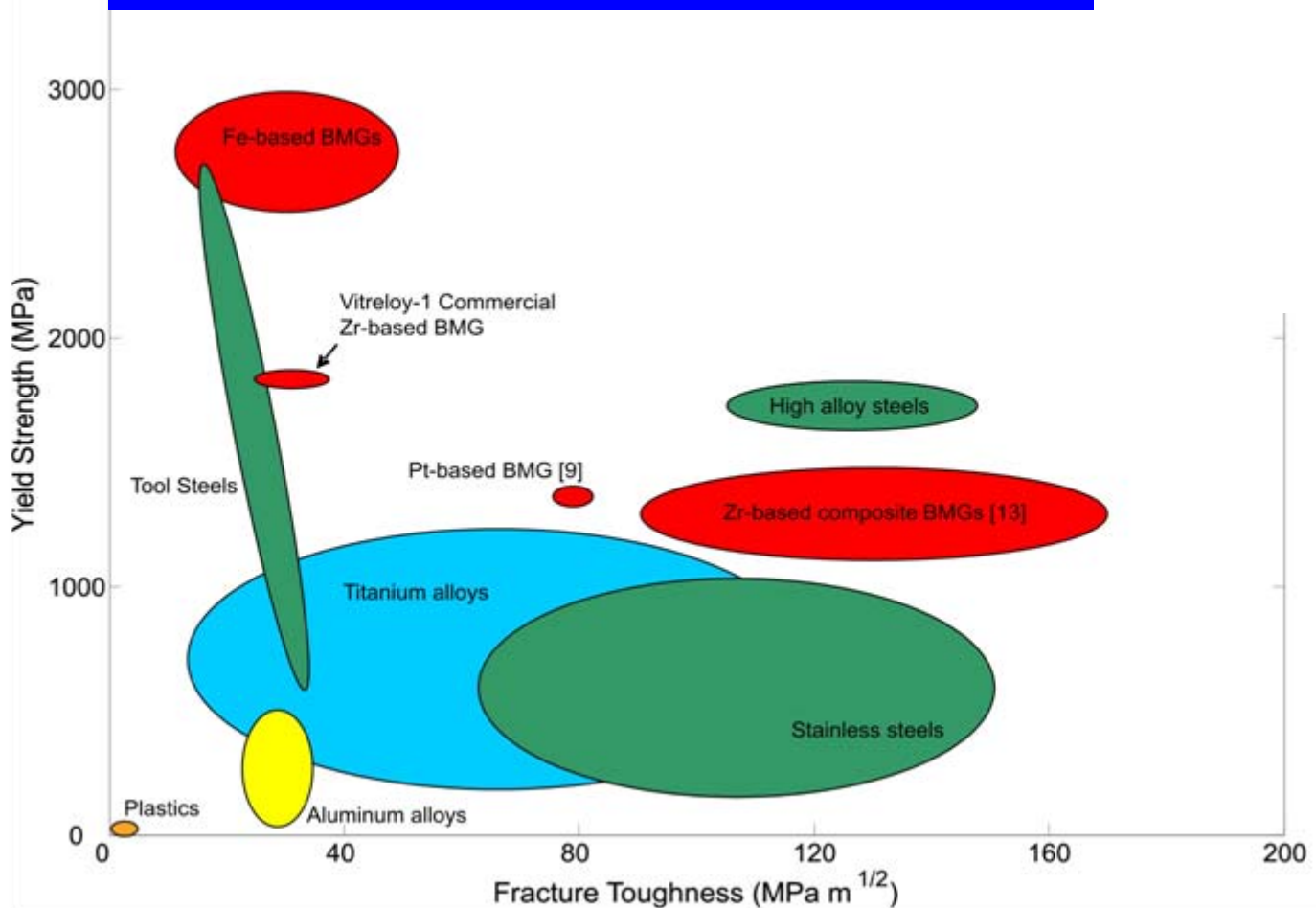
Medium	kW h / kg
Hydrogen	38
Gasoline	14
Dynamite	1.2
Carbon fiber flywheel	0.2
Steel flywheel	0.05
Lead-acid battery	0.04
Amorphous Fe spring	0.005
Steel spring	0.001

Mechanical properties at room temperature

Properties	$\text{Mg}_{60}\text{Cu}_{30}\text{Y}_{10}$	$\text{Zr}_{55}\text{Cu}_{20}\text{Ni}_{10}\text{Al}_{15}$	$\text{Zr}_{44}\text{Cu}_{40}\text{Al}_8\text{Ag}_8$	$\text{Zr}_{41.2}\text{Be}_{22.5}\text{Ti}_{13.8}\text{Cu}_1$ 2.5Ni_{10} Vitrelloy1	$\text{Zr}_{44}\text{Ti}_{11}\text{Cu}_{10}\text{Ni}_{10}\text{Be}_2$ 5 Vitrelloy 1B	$\text{Zr}_{58.5}\text{Cu}_{15.6}\text{Ni}_{12.8}$ $\text{Al}_{10.3}\text{Nb}_{2.8}$ Vitrelloy 106	$\text{Pt}_{57.5}\text{Cu}_{14.7}\text{Ni}_{5.3}\text{P}_{22.5}$	$\text{Fe}_{48}\text{Cr}_{15}\text{Mo}_{14}\text{Er}_2\text{C}_{15}\text{B}_6$
E, GPa	70	90	115 ⁷	96	96.7	87.3 ⁶	94.8	213
G, GPa	(25)	32		37.4	35.6	31.9 ⁶	33.3	81 ⁴
Yield strength, compression, MPa	850	1810 ²	1900	1900	1860	1820	1470	3750 ⁴
Plastic strain at fracture, compression	0.007%	0.2% ²	0.1 ⁷	0	0	0.5%	~20%	0
Fracture strength tension, MPa	ca. 600	1760 ¹		1800	1800	1200		
Plastic strain at fracture, tension	0%	0% ¹				0%		
Vickers hardness, GPa	4.3	5.5		5.4	5	5.9 ⁶	4.02	13 ⁵
Fracture toughness, MPa $\text{m}^{1/2}$		43 ³	17 ⁷	55			80	12.7 ± 4.1

1. $\text{Zr}_{52.5}\text{Ti}_5\text{Cu}_{17.9}\text{Ni}_{14.6}\text{Al}_{10}$ 2. $\text{Zr}_{58}\text{Ni}_{13.6}\text{Cu}_{18}\text{Al}_{10.4}$ 3. $\text{Zr}_{55}\text{Cu}_{30}\text{Ni}_5\text{Al}_{10}$ 4. $\text{Fe}_{49}\text{Cr}_{15}\text{Mo}_{14}\text{C}_{15}\text{B}_6\text{Er}_1$ 5. $\text{Fe}_{48}\text{Cr}_{15}\text{Mo}_{14}\text{C}_{15}\text{B}_6$ 6. $\text{Zr}_{57}\text{Nb}_5\text{Cu}_{15.4}\text{Ni}_{12.6}\text{Al}_{10}$ 7. $\text{Zr}_{48}\text{Cu}_{36}\text{Al}_8\text{Ag}_6$

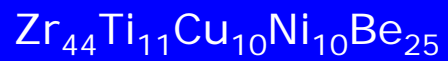
Comparison of mechanical properties for some BMGs with conventional alloys





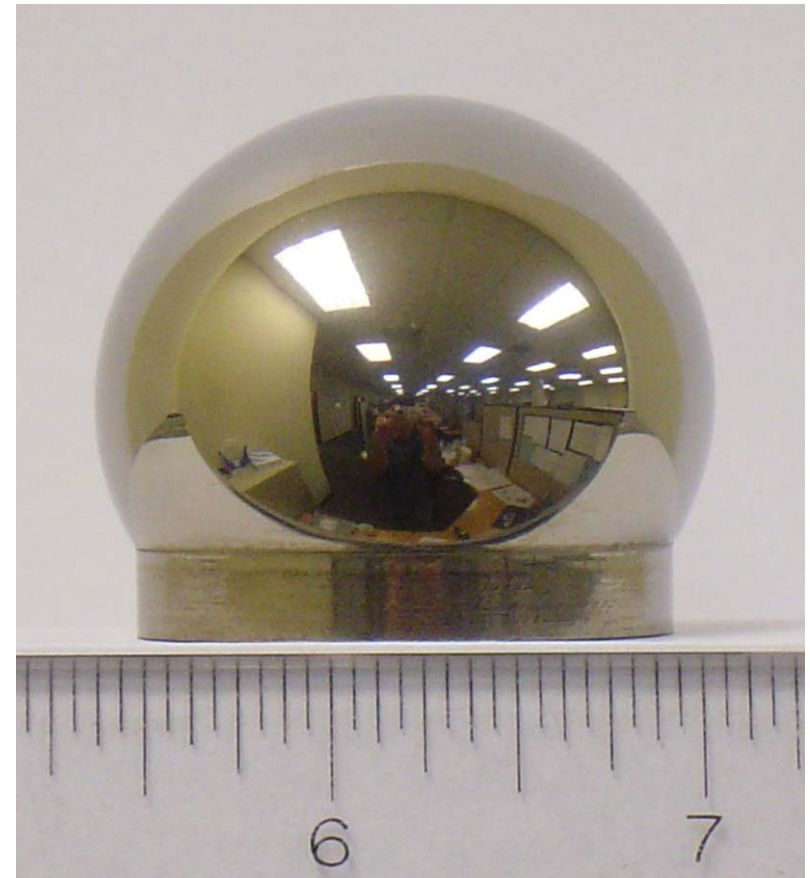
LIQUIDMETAL®

BLOW-MOLDING



$T=460^{\circ}\text{C}$, $t=100\text{ sec}$

Lung power, $\sim 10^4\text{ Pa}$




$T=460^{\circ}\text{C}$, $t=40\text{ sec}$

10^5 Pa , 400% strain

The unique characteristic of all BMG is the tremendous softening that occurs abruptly above T_g

- Microcomponent and microdevice shaping
- Faithful replication down to nm sizes (dependent on aspect ratio)
- Smooth surfaces

- Each BMG is an individual material with its own characteristics
- No BMG has all desirable characteristics: hard, strong, fatigue resistant, corrosion resistant, extremely formable, cheap, ...
- No known BMG exhibits plastic strain in unconstrained tension
- All BMG contain at least 3 elements, composition is not very flexible

Thanks for collaboration with colleagues at Risø 

and participants in the innovation consortium
MIKROMETAL:



Oticon

